

Catalytic Asymmetric Synthesis of Halenaquinone and Halenaquinol

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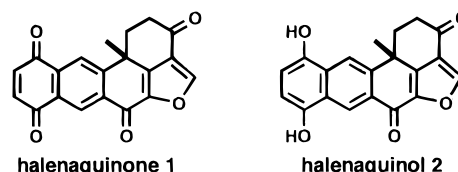
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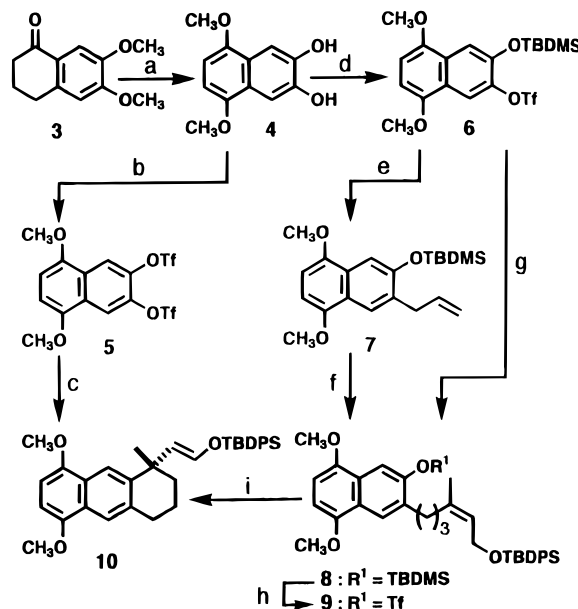
Halenaquinone (**1**) and halenaquinol (**2**), which have a benzylic quaternary carbon center, have been isolated from a variety of sea sponges (Chart 1).¹ These marine natural products have been shown to possess antibiotic, cardiotoxic, and protein tyrosine kinase inhibitory activity.² To date, only Harada and co-workers have succeeded in the total synthesis of **1** and **2** starting from optically pure Wieland–Miescher ketone.³ We report here the catalytic asymmetric synthesis of **1** and **2** starting from commercially available 6,7-dimethoxy-1-tetralone (**3**).^{4,5} This synthesis features the first use of a cascade Suzuki cross-coupling and an asymmetric Heck reaction as well as the one-pot construction of a pentacyclic ring system from a tricyclic ring system.

The catalytic asymmetric synthesis of the tricyclic key intermediate **10**, which has a benzylic quaternary carbon center, was achieved through three different routes (Scheme 1). The dihydroxynaphthalene derivative **4** was synthesized from **3** in 58% overall yield (five steps) as shown in Scheme 1, and was first converted to the ditriflate **5** (99%). To the best of our knowledge, there has been no previous example of a cascade Suzuki cross-coupling⁶ and an asymmetric Heck reaction.⁷ However, we considered that this quite challenging process could be possible under suitable reaction conditions to give optically active **10** in a single step. Therefore, alkylborane **12** was prepared,⁸ and after many attempts, we were very pleased to find that treatment of **5** with **12**⁹ (1.1 equiv), Pd(OAc)₂ (20 mol %), (*S*)-BINAP¹⁰ (40 mol %), and K₂CO₃ (6 equiv) in THF (0.063 M) at 60 °C for 42 h gave **10** with 85% ee in 20% yield.¹¹ The enantiomeric excess of **10** was determined by DAICEL CHIRAL-CEL OD (hexane:2-propanol, 9:1) using **15** readily derived from **10**. The absolute configuration of **10** was unequivocally determined from the successful conversion of **10** to natural halenaquinone and halenaquinol. Al-

Chart 1. Structure of **1** and **2**



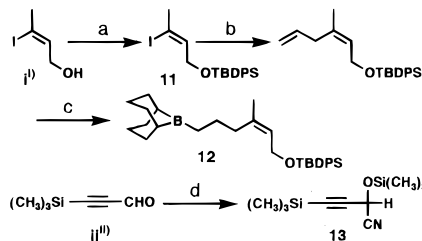
Scheme 1. Catalytic Asymmetric Synthesis of **10**^a



^a Reaction conditions: (a) (1) BBr₃ (2.1 equiv), CH₂Cl₂, -78 °C to rt, (2) BnBr (2.0 equiv), K₂CO₃, *n*-Bu₄Ni, DMF, 60 °C, (3) CrO₃ (5 equiv), AcOH–H₂O, 0 °C to rt, (4) KHMDS (3 equiv), THF, -78 °C, then MeI (6 equiv), -78 °C to rt, (5) H₂ (1 atm), Pd–C, AcOEt, rt (five steps, 58%); (b) Tf₂O (3 equiv), pyridine, CH₂Cl₂, -78 °C to rt (99%); (c) **12** (1.1 equiv), Pd(OAc)₂ (20 mol %), (*S*)-BINAP (40 mol %), K₂CO₃, THF, 60 °C (85% ee, 20%); (d) (1) TBDMSCl (1.1 equiv), Et₃N, CH₂Cl₂, 0 °C, (2) Tf₂O (1.3 equiv), Et₃N, CH₂Cl₂, -78 °C to rt (two steps, 85%); (e) CH₂=CHCH₂MgBr (5 equiv), PdCl₂(dppf)·CH₂Cl₂ (9 mol %), Et₂O, -78 °C to rt (quant); (f) (1) 9-BBN (2.1 equiv), THF, 0 °C to rt, (2) **11** (1.5 equiv), PdCl₂(dppf)·CH₂Cl₂ (5 mol %), K₃PO₄·*n*H₂O, THF, 50 °C (90%); (g) **12** (1.3 equiv), PdCl₂(dppf)·CH₂Cl₂ (10 mol %), K₂CO₃, THF, 50 °C (61%); (h) (1) *n*-Bu₄NF (1.0 equiv), THF, 0 °C, (2) Tf₂O (1.3 equiv), Et₃N, CH₂Cl₂, -78 °C to rt (two steps, 69%); (i) Pd(OAc)₂ (10 mol %), (*S*)-BINAP (20 mol %), K₂CO₃, THF, 60 °C (87% ee, 78%).

though the chemical yield is still unsatisfactory, this is the first example of a cascade Suzuki cross-coupling and

(8) Compounds **11**–**13** were prepared as follows (II) Cochrane, J. S.; Hanson, J. R. *J. Chem. Soc., Perkin Trans. 1* **1972**, 3, 361. (II) Kruithof, K. J. H.; Schmits, R. F.; Klumpp, G. W. *Tetrahedron* **1983**, 39, 3073.



(a) TBDPSCl, Et₃N, dimethylaminopyridine, CH₂Cl₂ (quant); (b) CH₂=CHCH₂MgBr, PdCl₂(dppf)·CH₂Cl₂ (5 mol %), Et₂O (76%); (c) 9-BBN (1.0 eq.), THF; (d) TMSMgCl, ZnI₂ (cat.) (91%)

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(4) Purchased from Aldrich Chemical Co., Inc.

(5) For a synthetic approach to **1** and **2** using a Heck reaction, see: Cristofoli, W. A.; Keay, B. A. *Synlett* **1994**, 625.

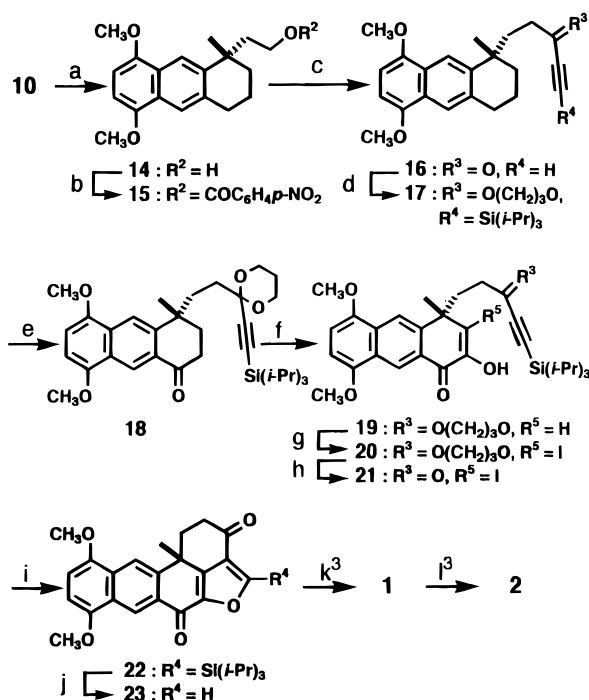
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an asymmetric Heck reaction. Alternatively, the benzylic quaternary carbon **10** was also constructed through a multistep sequence of reactions. **4** was converted to the silyl ether, which underwent trifluoromethanesulfonylation to give **6** in 85% overall yield. The Pd(0)-catalyzed cross-coupling reaction gave **7** in a quantitative yield, which underwent hydroboration followed by a Suzuki cross-coupling reaction using **11**⁸ to give **8** (90%). Silyl ether **8** was also constructed by a Suzuki cross-coupling reaction using **6** and trialkylborane **12** (61%). **8** was then converted to the requisite triflate **9** (69%) for an asymmetric Heck reaction in a two-step sequence of reactions. Treatment of **9** with Pd(OAc)₂ (10 mol %), (*S*)-BINAP (20 mol %), and K₂CO₃ (3 equiv) in THF at 60 °C for 22 h gave **10** with 87% ee in 78% yield.

Having developed a catalytic asymmetric synthesis for the key intermediate **10**, which has a benzylic quaternary carbon center, we then examined the conversion of **10** to natural halenaquinone and halenaquinol (Scheme 2). We expected that the alkenyl iodide **21** would be a reasonable synthetic intermediate for constructing the pentacyclic ring system in a one-pot reaction.¹² Therefore, **10** was first converted to alcohol **14** in a two-step sequence of reactions (93%). Transformation of **14** to the triflate followed by reaction with the acyl anion equivalent **13**⁸ and then 2% NaF gave ketone **16** (68% yield). After protection of the carbonyl functionality as an acetal (98%), and of the ethynyl functionality with a triisopropylsilyl group (98%), **17** underwent benzylic oxidation to give **18** (96%). Exposure of **18** to O₂ (1 atm) in the presence of KO-*t*-Bu (5 equiv) in *tert*-butyl alcohol gave enol **19** (79%). Treatment of **19** with NaI (10 equiv) and CuSO₄·5H₂O (10 equiv) in aqueous methanol gave the requisite alkenyl iodide **20** quite efficiently (97%),¹³ and exposure of **20** to *p*-toluenesulfonic acid in aqueous acetone furnished **21** (98%). We were very pleased to find that reaction of **21** with Pd₂(dba)₃·CHCl₃ (0.28 molar equiv) and K₂CO₃ (5 equiv) in DMF at room temperature for 8 h gave the desired pentacycle **22** in a single step (72%). The pentacyclic intermediate **22** was subjected to desilylation, which gave **23** in 83% yield: [α]_D²³ +123.7 (*c* 0.335, CH₂Cl₂, 87% ee). Finally, Harada's synthetic

Scheme 2. Synthesis of **1** and **2** from **10**^a



^a Reaction conditions: (a) (1) *n*-Bu₄NF (2 equiv), AcOH (3 equiv), THF, 0 °C to rt, (2) NaBH₄ (5 equiv), MeOH, 0 °C to rt (two steps, 93%); (b) 4-nitrobenzoyl chloride (1.1 equiv), Et₃N, 4-(dimethylamino)pyridine, CH₂Cl₂, 0 °C to rt (96%); (c) (1) Tf₂O (1.2 equiv), pyridine, CH₂Cl₂, -78 °C to rt, (2) LDA/**13** (1.5 equiv), THF, -78 °C, then 2% NaF (two steps, 68%); (d) (1) HO(CH₂)₃OH (10 equiv), TsOH·H₂O, benzene, reflux (98%), (2) *n*-BuLi (2 equiv), THF, -78 to -50 °C, then TIPSCl (2 equiv), -78 °C to rt (98%); (e) DDQ (3 equiv), CH₂Cl₂, H₂O, rt (96%); (f) O₂ (1 atm), KO-*t*-Bu (5 equiv), *t*-BuOH, 35 °C (79%); (g) NaI (10 equiv), CuSO₄·5H₂O (10 equiv), MeOH, H₂O, rt (97%); (h) TsOH·H₂O, acetone, H₂O, 60 °C (98%); (i) Pd₂(dba)₃·CHCl₃ (0.28 equiv), K₂CO₃, DMF (0.003 M), rt (72%); (j) *n*-Bu₄NF (16 equiv), AcOH (24 equiv), THF, MeCN, 60 °C (83%); (k) CAN (5 equiv), MeOH, H₂O, 0 °C (45%); (l) Na₂S₂O₄ (20 equiv), acetone, H₂O, 0 °C (quant).

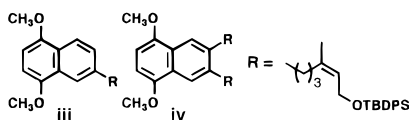
intermediate **23**³ was readily converted to halenaquinone (**1**) and halenaquinol (**2**).

In conclusion, we have developed a catalytic asymmetric synthesis of **1** and **2** (87% ee) that incorporates the first example of a cascade Suzuki cross-coupling–asymmetric Heck reaction as well as a single-step process promoted by Pd(0) for constructing a pentacyclic ring system from a tricyclic ring system.

Supporting Information Available: Details of the preparation and characterization of **1**, **2**, **5**, **6**, **8–10**, **14**, **16**, and **18–23** (29 pages).

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(11) The byproducts are **iii** and **iv**.



(12) The related reaction for the synthesis of indole and furan skeletons has been previously reported. See: Larock, R. C.; Yum, E. K.; Doty, M. J.; Sham, K. K. C. *J. Org. Chem.* **1995**, *60*, 3270.

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